

**II B.Tech II Semester Supplementary Examinations,
November/December 2005
MATERIAL SCIENCE FOR CHEMICAL ENGINEERING
(Chemical Engineering)**

Time: 3 hours**Max Marks: 80**

**Answer any FIVE Questions
All Questions carry equal marks**

1. (a) State the different types of bonds and their characteristics.
(b) Name four types of atomic bonds observed in the materials. Describe them in detail by giving examples. [8+8]
2. Sodium and chlorine weigh 22.997 and 35.457gm/atomic weight respectively. If the density is 2.165 gm/cm^3 . Calculate the dimensions of unit cell of NaCl. [16]
3. (a) What is a solid solution?
(b) Discuss the similarities and differences between substitutional and interstitial solid solution. [6+10]
4. The energy stored by strain hardening in a metal is $4 \times 10^4 \text{ cal/m}^3$. Given that the shear modulus is $4 \times 10^{10} \text{ N/m}^2$, $b = 2.8 \times 10^{-10} \text{ m}$ and $\nu = \frac{1}{3}$, calculate the dislocation density. Assume that the edge and screw dislocation densities are equal. [16]
5. Explain the differences between:
 - (a) Hardness and brittleness
 - (b) Strength and stiffness
 - (c) Ductility and malleability
 - (d) Elasticity and plasticity [4x4=16]
6. Sketch time-strain curve for anelastic behavior and compare it with the time-strain curve in the Voigt-Kelvin element? What do you deduce from this? [16]
7. What is fracture? State the causes of fracture in a material? [16]
8. Define grain boundary and account for its character. [16]

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1. How many atoms are there per square millimeter on:
 - (a) (100) Plane of copper
 - (b) (110) Plane
 - (c) (111) Plane? [5+5+6]
2. Write about closed packed structures. [16]
3. What is meant by crystal imperfections? Classify them in order of their geometry. [16]
4. Consider a hypothetical element whose crystal structure consists of atoms on the points of a simple-cubic lattice. Make a sketch of crystal structure for a left-hand screw orientation. [16]
5. Explain the differences between :
 - (a) Elastic limit and proportional limit
 - (b) Yield point and yield strength
 - (c) Toughness and resilience
 - (d) Endurance limit and fatigue strength [4x4=16]
6. Sketch time-strain curve for anelastic behavior and compare it with the time-strain curve in the Voigt-Kelvin element? What do you deduce from this? [16]
7. How would you distinguish between a ductile and a brittle fracture? Why is the fracture strength of real materials lower than ideal breaking strength? [16]
8. A liquid containing 90% Ni and 10% Cu is cooled relatively quickly the bottom of the container upward. Describe the compositional differences in the final solid casting. [16]

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1. (a) Sketch the five, two dimensional (plane) lattices.
(b) Explain why each is a distinct distribution of points in two dimension.
(c) Why is “entered squar” not a separate plane lattice. [10+3+3]
2. Calculate the atomic density (number of atoms for unit area) in (111),(110),(100) planes of copper (F.C.C.) with the lattice parameter of 3.61 \AA . Can you pack atoms more closely than in (111) plane? [16]
3. State the effects of crystal imperfections in materials. [16]
4. (a) Briefly discuss about the interactions of dislocations?
(b) Give simple rules for interaction between two edge dislocations? [8+8]
5. Relatively brittle materials can often be successfully deformed by special processes in which two or three principal stresses are applied simultaneously, one a tensile stress and the other a compressive stress of comparable magnitude. Using the following equation, show that plastic flow can occur even if each of these stresses is appreciable below the yield stress σ_0 at which plastic flow begins when only a single tensile stress is applied.
$$\bar{\sigma} = \left\{ \frac{1}{2} [(\sigma_I - \sigma_{II})^2 + (\sigma_{II} - \sigma_{III})^2 + (\sigma_{III} - \sigma_I)^2] \right\}^{1/2}$$
[16]
6. What is understood by viscoelastic deformation? State the mechanical models to demonstrate viscoelastic behavior of materials? Name the materials which exhibit viscoelastic behavior? [16]
7. (a) Give procedure for shortening the relaxation time of a plastic.
(b) Cite examples where long relaxation times are desirable? [10+6]
8. How much ferrite, austenite and cementite by weight are presenting 0.1 kg of an Fe-C alloy containing 0.5 wt% C at
(a) 1000°C
(b) 724°C and room temperature? [16]

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1. Sketch a (110) plane in a unit cell of cubic crystal. Show all the (111) directions that lie on this plane, giving Miller indices of each one of them. [16]
2. X-rays with a wave length of 1.54\AA are used to calculate the spacing of (200) planes in aluminium. The Brag angle for this reflection 22.4° . What is the size of the unit cell of the aluminium crystal? [16]
3. Give a sketch of stacking fault in a fcc structure and describe about stacking fault. [16]
4. A two phase material has precipitate particles of radius $1.25 \times 10^{-8} \text{ m}$ and of concentration 10^{20} m^{-3} . The energy of the grain boundary between the particles and the surrounding material is 0.5 Jm^{-2} . After the heat treatment at an elevated temperature the particles were found to have coarsened to $3 \times 10^{-8} \text{ m}$ radius and their concentration decreased to $7.2 \times 10^{18} \text{ m}^{-3}$. What is the reduction in the total grain boundary energy due to the coarsening? [16]
5. Explain the differences between :
 - (a) Elastic limit and proportional limit
 - (b) Yield point and yield strength
 - (c) Toughness and resilience
 - (d) Endurance limit and fatigue strength [4x4=16]
6. In dislocation density in a copper sample is increased by cold working from 10^9 m^{-3} to 10^{13} m^{-3} calculate the free energy change during recrystallization. [16]
7. What is fracture? State the causes of fracture in a material? [16]
8. Define grain boundary and account for its character. [16]
